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Twinning in barium titanate crystals*. By E. A. D. WHITE, *Research Laboratories of the General Electric Company Limited, Wembley, England*

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Single crystals of barium titanate have been grown in these laboratories by a technique based on the method due to Remeika (1954). The investigation of the domain structure and electrical properties of barium titanate crystals has received considerable attention (e.g. Merz, 1952, 1954), but few details have been published of the interesting crystallographic features displayed by crystals grown from solution in potassium fluoride.

The so-called 'butterfly twin' crystals obtained consist of two triangular plates joined along the hypotenuse at an acute angle. A narrow, well developed face is present along the junction of the plates (Fig. 1(a)). Occasionally

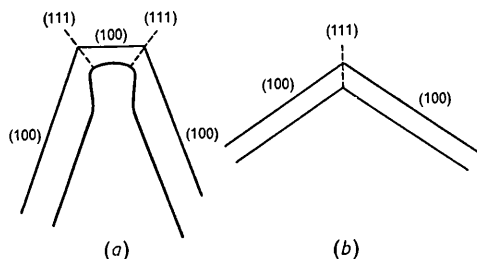


Fig. 1. Sections through the junction of BaTiO_3 twinned crystals. (a) Acute-angled 'butterfly twin'; (b) obtuse-angled crystal.

crystals are found which have the plates joined at an obtuse angle. Crystals of the latter type possess a sharp edge along the junction (Fig. 1(b)). It may be noted in Fig. 1(a) that in the acute-angled crystals tapering of the thickness of the plates occurs near to the junction. This is presumably due to restricted access of solution to the narrow space between the two crystal plates.

Measurement of the angle between the wings for a large number of crystals showed it to be approximately constant. The small variations observed may be ascribed to differences in the axial ratio of the structure cell, together with variations in the ratio of a to c domains (Forsbergh, 1949) at the junction.

The mean value obtained is consistent with a (114) twinning plane. Consideration of the atomic arrangement for the (114) plane, however, did not suggest a definite

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Table 1. Calculated and observed angles between the 'wings' of barium titanate twin crystals

Type of crystal	Calculated angle	Observed angle	
Acute-angled twins	Entirely a domain ($c/a = 1.01$)	$39^\circ 18'$	—
	Entirely c domain ($c/a = 1.01$)	$38^\circ 34'$	—
	Equal a and c domains present (equivalent to $c/a = 1$)	$38^\circ 58'$	—
	Mean observed value	—	$38^\circ 49'$
Obtuse-angled twins	Equal a and c domains present (equivalent to $c/a = 1$)	$109^\circ 29'$	—
	Mean observed value	—	$109^\circ 31'$

reason for such twinning. Furthermore, the small face at the junction of the acute-angled crystals would have the indices 221 for a (114) twinning plane. Such a face should rapidly disappear during growth owing to differential growth rates between (221) and (100).

A close study of many crystals has shown that, in fact, two twinning planes are present in each 'butterfly twin', both of which are (111) planes. The observed narrow face at the junction is thus a (100) face, and is consequently retained throughout the development of the crystal. The obtuse-angled crystals possess only a single (111) twinning plane and thus have a sharp junction.

The dendritic type of growth, which is readily obtained from potassium fluoride solution, also shows (111) twinning. In this case sheets of crystals are formed in which either tabular crystals or pyramids project on both sides and in all possible orientations from the common (111) plane.

Twinning in barium titanate thus conforms to established twinning laws for cubic crystals. The 'butterfly twin' habit is evidently a result of the prevalence of twinning, although the reasons for the precise form of this twin are not yet clear.

References

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On the structure of the crystal form A' of n -pentadecanoic acid. By ERIK VON SYDOW, *Institute of Chemistry, University of Uppsala, Uppsala, Sweden*

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The structure described by von Sydow (1954) has been refined using difference syntheses. A temperature factor $\exp[-B \sin^2 \theta/\lambda^2]$, with $B = 3.0 \text{ \AA}^2$, has been applied.

The coordinates of the hydrogen atoms were derived as in von Sydow (1954), and these values agree very well with the maxima obtained in the ($F_o - F_c$) synthesis